

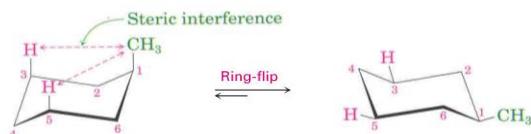
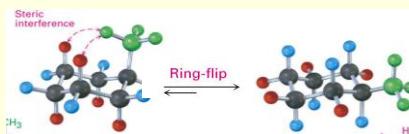
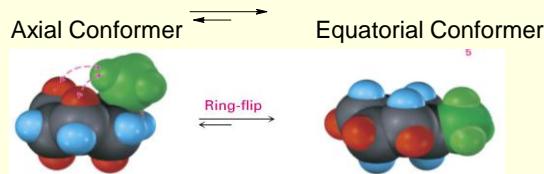
الهكسان الحلقي ذو البديل الواحد

Mono substituted cyclohexane

1

Monosubstituted Cyclohexanes

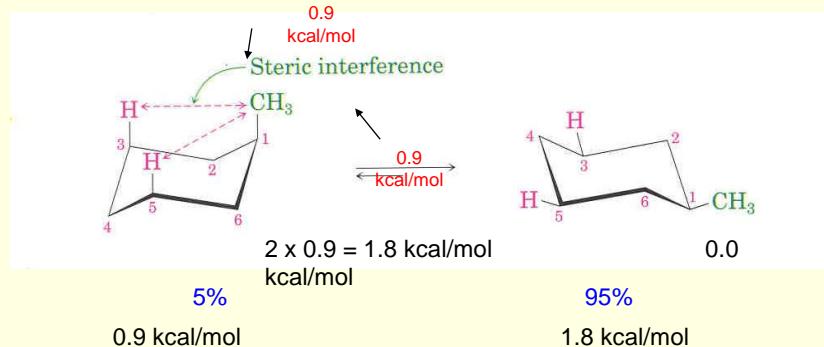
Methylcyclohexane



4-2

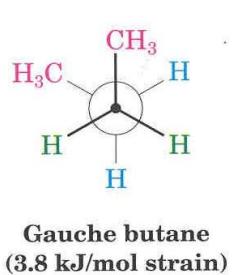
The equatorial conformer is more stable by 1.8 kcal/mol due to a steric interaction between the axial methyl and two axial hydrogens. These interactions are termed 1,3-diaxial interactions.

سؤال / أيهما أكثر ثباتاً الوضع الأفقي أم الوضع العمودي للمركب $\text{1-methylcyclohexane}$

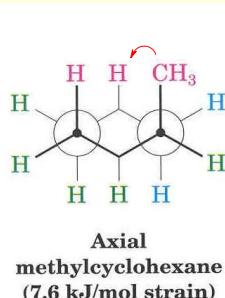


4-3

These 1,3-diaxial interactions are really gauche interactions in disguise! Remember that a gauche interaction for butane is worth 0.9 kcal/mol.

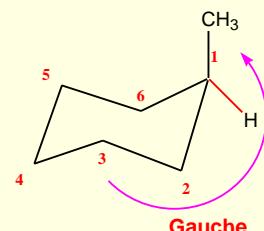


0.9 kcal/mol

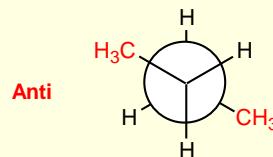


Axial methylcyclohexane (7.6 kJ/mol strain)

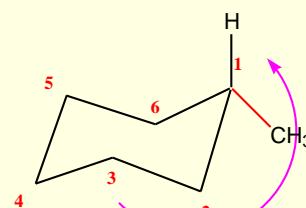
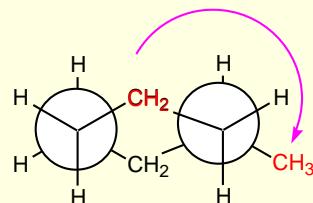
4-4



1.8 kcal/mol



الوضع المتصاد للبيوتان



الوضع الافقى
1-
methylcyclohexane

4-5